PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



DITERNATIONAL APPLICATION PUBLISI		UNDER THE PATENT COOPERATION TREATY (PCT)
(51) International Patent Classification 6: A61K 9/12, 31/135, 31/46	A1	(11) International Publication Number: WO 99/65464 (43) International Publication Date: 23 December 1999 (23.12.99
(21) International Application Number: PCT/US (22) International Filing Date: 8 June 1999 (HR, HU, ID, IL, IN, JP, KR, KZ, LT, LV, MX, NU, NZ
(30) Priority Data: 198 27 178.6 198 42 963.0 19 September 1998 (18.06.98) 19 September 1998 (19.09.9		DE Published With international search report.
(71) Applicant: BOEHRINGER INGELHEIM PHARM CALS, INC. [US/US]; 900 Ridgebury Road, Ridge 06877 (US).	ACEU efield, (7- 27
(72) Inventors: McNAMARA, Daniel, P.; 30 Newton Waterbury, CT 06708 (US). DeSTEFANO, Georg Greenknoll Drive, Brookfield, CT 06804 (US).	n Terra ge, A.;	ze, 13
(74) Agents: RAYMOND, Robert et al.; Boehringer Pharmaceuticals, Inc., 900 Ridgebury Road, P.O. Ridgefield, CT 06877 (US).	Ingelhe Box 3	im
(54) Title: PHARMACEUTICAL FORMULATIONS FO	OR AE	ROSOLS WITH TWO OR MORE ACTIVE SUBSTANCES
administration by inhalation or by nasal route. Specific	cally, u	mulations for aerosols with at least two or more active substances in invention relates to pharmaceutical preparations for propellant-driving pellant, which contain a combination of active substance of two or more than the combination of active substance in the combination of the combination of active substance in the combination of t
·		;

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

	-						
AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Amenia	n	Finland	LT	Lithuania	SK	Slovakia
	Amenia	FR	France	LU	Luxembourg	SN	Senegal
AT	Australia	GA	Gahon	LV	Latvia	SZ	Swaziland
AU	***************************************	GB	United Kingdom	MC	Monaco	TD	Chad
AZ	Azerbaijan	GB	Georgia	MD	Republic of Moldova	TG	Togo
BA	Bosnia and Herzegovina	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BB	Darbados		Guinea	MK	The former Yugoslav	TM	Turkmenistan
BE	Belgium	GN	Greece		Republic of Macedonia	TR	Turkey
BP	Burkina Paso	GR		ML	Mali	77	Tripidad and Tobago
BC	Bulgaria	HU	Hungary	MN	Mongolia	UA	Ukraine
BJ	Benis	(B	Ireland	MR	Mauritania	UG	Uganda ·
BR	Brazil	IL	Israel	MW	Malawi	US	United States of Americ
BY	Belanus	IS	Iceland			UZ	Uzbekistan
CA	Canada	п	Italy	MX	Mexico	VN	Viet Nam
CF	Central African Republic	JP	Japan	NB	Niger	YU	
CG	Congo	KB	Kenya	NL	Netherlands		Yugaslavia Zimbabwa
CH	Switzerland	KG	Kyrgyzsian	NO	Norway	ZW	Zimonowe
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
CM	Cameroon		Republic of Korea	PL	Poland		
CN	China	KR	Republic of Korca	PT	Portugal		
CU	Cuba	ΚZ	Kuzakstan	RO	Romania		
cz	Czech Republic	LC	Saim Lucia	RU	Russian Federation		
DE,	Germany	u	Licchtenstein	SD	Sudan		
DK	Denmark	LK	Sri Lanka	SE	Sweden		
BE	Estonia	LR	Liberia	SG	Singapore		

WO 99/65464 PCT/US99/12785

Prosecution application

5 Pharmaceutical Formulations for Aerosols with two or more Active Substances

The present invention relates to new pharmaceutical formulations for aerosols with at least two or more active substances for use by inhalation or by the nasal route.

State of the art

10

25

In propellant-driven metered dose inhalers (MDI) the
active substances can be formulated as solutions or
suspensions. The vast majority of aerosol formulations
for MDI's are prepared as suspensions, especially if the
preparation contains more than one active substance.
Formulations in the form of solutions are used only to a
limited extent. In these cases, the formulations normally
contain only one active substance.

As a rule, in a suspension, the chemical stability of the active substances is noticeably higher than in a solution. Additionally, in a suspension the active substance can be more highly concentrated than in a solution, with the result that suspension type formulation enable higher doses to be administered.

A major disadvantage of suspension-formulations is the fact that over time (e.g. during storage) the suspended particles clump together to form bigger, more or less stable agglomerates or form loose flakes, sediments or floating layers, or in the worst case, particle growth, which significantly impairs the pharmaceutical quality of the product. The size of the particles formed or the

speed of particle growth are influenced by the solubility features of the liquid phase. Thus, ingress of humidity during storage or a desired increase in polarity, e.g. achieved by adding co-solvents, can have a devastating effect on the quality of the medical end product, particularly if the suspended particles have polar structure elements. The suspension can be physically stabilised by the addition of surfactants, by reducing the harmful effects of moisture and/or particle growth so that suspended particles can be held in suspension for longer.

Natural solution-type formulations are not affected by the problems of increasing particle size or de-mixing processes such as sedimentation or flocculation. However, in this case there is a serious risk of chemical 15 degradation. A further disadvantage is the fact that the limited solubility of the ingredients can prevent administration in high doses. In the past, the chlorofluorohydrocarbons TG 11 (trichlorofluoromethane), TG 12 (dichlorodifluoromethane) and TG 114 20 (dichlorotetrafluoroethane) have proved particularly suitable as solvents. The solubility of the ingredients can be increased by the addition of co-solvents. In addition, it is usually necessary to take additional measures to chemically stabilise the dissolved components. 25

Up till now, CFCs such as the above-mentioned TG 11, for example, have often been used as propellants. However, since CFCs have been linked with the destruction of the ozone layer, their manufacture and use are being phased out. The intention is to replace them with special fluorohydrocarbons (HFC) which are less destructive to the ozone layer but have completely different solubility features. The toxicological profile and physico-chemical properties such as the steam pressure, for example, determine which HFCs are suitable for MDIs. The most

promising representatives at present are TG 134a (1,1,2,2-tetrafluoroethane) and TG 227 (1,1,1,2,3,3,3-heptafluoropropane).

For inhalative treatment it may be desirable to have aerosol formulations with two or more active substances. In such cases the active substances are formulated in the necessary concentrations as solutions or suspensions, frequently giving rise to problems regarding the chemical stability of the individual substances or the degree of concentration which can be attained. Major problems are encountered if one of the active substances cannot be suspended or is unstable in a suspension-type formulation of this kind or if one of the active substances is chemically unstable or will not dissolve in a solution-type formulation of this kind, particularly when HFC is used as the propellant.

It is therefore one object of the present invention to

20 develop a formulation for metering aerosols having two or
more active substances which overcomes the above-mentioned
disadvantages.

Description of the invention

25

Surprisingly, it has been found that a plurality of active substances can be formulated as a solution and a suspension combined in one formulation.

The invention relates to stable aerosol formulations with fluorohydrocarbons as propellants, particularly TG 134a and/or TG 227, consisting of two or more active substances, wherein at least one active substance is formulated as a solution and at least one active substance is formulated as a suspension. The pharmaceutical preparation according to the invention is used for

inhalative treatment, particularly for treating diseases of the pharynx and respiratory tract, e.g. asthmatic diseases and COPD.

5 Detailed description of the invention

In one embodiment a medicinally useful combination of two or more active substances is used, containing beclometasone, budesonide, cromoglycinic acid, fenoterol, flunisolide, fluticasone, ipratropium bromide, nedocromil, orciprenaline, oxitropium bromide, reproterol, salbutamol (albuterol), salmeterol, terbutalin, N-[[2,2-dimethyl-4-(2-oxo-2H-pyridin-1-yl)-6-trifluoromethyl-2H-1-benzopyran-3-yl]methyl]-N-hydroxy-acetamide, the esters, salts and/or solvates thereof. Which of the above-mentioned active substances is formulated as a solution and which as a suspension in the preparation according to the invention depends on the particular combinations of active substance and can be determined relatively quickly by solution and suspension trials.

In a preferred embodiment, one or more of the following active substances are suspended: budesonide, cromoglycinic acid, nedocromil, reproterol and/or salbutamol (albuterol) or the esters, salts and/or solvates derived from these compounds and one or more of the following substances are dissolved: beclomethasone, fenoterol, ipratropium bromide, orciprenaline and/or oxitropium bromide, N-[[2,2-dimethyl-4-(2-oxo-2H-pyridin-1-yl)-6-trifluoromethyl-2H-1-benzopyran-3-yl]methyl]-N-hydroxy-acetamide or the esters, salts and/or solvates derived from these compounds. Embodiments having two different active substances are preferred.

. 35 A particularly preferred embodiment contains dissolved ipratropium bromide, particularly combined with salbutamol

sulphate (albuterol sulphate) as the suspended active substance.

In all the embodiments, the active substances are used in a therapeutically effective quantity, i.e. in a quantity that can induce a successful treatment. The concentration of the active substances and the volume per stroke of spray are adjusted in such a way that the quantity of active substance which is medically necessary or recommended is released by a single spray or by a few sprays.

One embodiment relates to formulations in which the suspended particles are stabilised by the addition of surfactant substances (surfactants) or other suspension-stabilising agents to stabilise the suspended particles against physical changes. The benefit of this is that the particle size will remain pharmaceutically acceptable even over lengthy periods, e.g. during storage. Preferred particle sizes are up to 20 µm, whilst particularly preferred particle sizes are between 5 and 15 µm, best of all not exceeding 10 µm. The advantage of these particle sizes is that the particles are small enough to penetrate deeply into the lungs but not so small as to be breathed out again with the exchanged air.

Suitable surfactants and suspension-stabilising agents include all pharmacologically acceptable substances which have a lipophilic hydrocarbon group and one or more functional hydrophilic groups, especially $C_{5\cdot 20}$ fatty alcohols, $C_{5\cdot 20}$ fatty acids, $C_{5\cdot 20}$ fatty acid esters, lecithin, glycerides, propyleneglycol esters, polyoxyethylenes, polysorbates, sorbitan esters and/or carbohydrates. $C_{5\cdot 20}$ fatty acids, propyleneglycol diesters and/or triglycerides and/or sorbitans of the $C_{5\cdot 20}$ fatty acids are preferred, whilst oleic acid and sorbitan mono-,

di- or trioleates are particularly preferred.

Alternatively, toxicologically and pharmaceutically acceptable polymers and block-polymers can be used as suspension-stabilising agents. The surfactants used are either non-fluorinated or partially fluorinated or perfluorinated, the term fluorinated referring to the exchange of hydrogen radicals bound to the carbon for fluorine radicals. The quantity of surfactant may be up to 1:1 based on the proportion by weight of the suspended active substances; amounts of 0.0001:1 to 0.5:1 are preferred, whilst amounts of from 0.0001:1 to 0.25:1 are particularly preferred.

A further advantage of the above surfactants is that they can also be used as valve lubricants. Therefore, one embodiment relates to formulations in which said surfactants are added as valve lubricants.

In another embodiment the solubility of at least one
active substance to be dissolved is increased by the
addition of one or more co-solvents. This has the
advantage that the active substance or substances to be
dissolved can be formulated in higher concentrations. The
addition of co-solvent must not exceed the critical
threshold of polarity of the liquid phase at which one of
the disadvantages described above begins to affect the
suspended particles of active substance.

Suitable co-solvents are pharmacologically acceptable

alcohols such as ethanol, esters or water or mixtures
thereof; ethanol is preferred. The concentration of the
co-solvent in relation to the total formulation may be
from 0.0001 to 50 wt.-%, preferably 0.0001 to 25 wt.-%.
In another embodiment a concentration of 0.0001 to

10 wt.-% is preferred whilst particularly preferred
embodiments are those wherein just enough alcohol is added

to dissolve the active substance which has to be dissolved.

In another embodiment, other common propellants are added to the HFC propellant. These added propellants may be, beside other HFCs, saturated lower hydrocarbons such as propane, butane, isobutane or pentane provided that the mixture is pharmacologically acceptable.

- In one embodiment, stabilisers are added to the formulation, with a beneficial effect on the pharmaceutical stability of the active substances even over lengthy periods, e.g. during storage. In the context of the invention, stabilisers denotes those substances
- which prolong the durability and usability of the pharmaceutical preparation by preventing or delaying chemical changes in the individual ingredients, particularly the active substances, e.g. caused by subsequent reactions or degradation, or those which
- prevent biological contamination. Stabilisers which are preferred for this purpose are those which influence the pH of the liquid phase, such as acids and/or the salts thereof, particularly suitable substances are hydrochloric acid, sulphuric acid, nitric acid, phosphoric acid,
- ascorbic acid, citric acid and the salts thereof. In addition, preferred bactericides, fungicides etc. are benzalkonium chloride or ethylene diamine tetraacetate. Citric acid is most preferred. The concentration of the stabilisers may be up to 1000 ppm, preferably up to 100 ppm and most preferably 20 to 40 ppm.
- One particularly preferred embodiment comprises suspended salbutamol sulphate (albuterol sulphate), dissolved ipratropium bromide, ethanol as co-solvent and citric acid

as stabiliser.

Examples

Example 1

In a solution of liquefied 89.96 g (1 mol, 89.71 wt.-%) of TG 134a and 10.03 g (218 mmol, 10.00% by weight) of ethanol are dissolved 37 mg (0.09 mmol, 0.037 wt.-%) of ipratropium bromide and 4 mg (20 μmol, 0.004% by weight) of citric acid and 210.5 mg (0.88 mmol, 0.21% by weight) of salbutamol sulphate (albuterol sulphate) are suspended together with 0.05% by weight of surfactant (e.g. 50 mg (177 mmol) of oleic acid).

Example 2

15

Analogous to Example 1 using TG 227 as the propellant gas instead of TG 134a.

Example 3

20

Disodium chromoglycate is suspended in liquefied P134 and a small amount of ethanol and fenoterol hydrobromide is dissolved therein.

25 Example 4

Analogous to Example 3 using TG 227 as propellant gas instead of TG 134a.

15

25

Patent Claims

- Pharmaceutical preparation for propellant driven metered dose inhalers having a fluorohydrocarbon
 (HFC) as propellant, which contain a combination of two or more active substances characterised in that at least one active substance is present in dissolved form by the use of a co-solvent together with at least one other active substance in the form of suspended particles.
 - Pharmaceutical preparation according to claim 1, characterised in that the combination of active substances consists of two active substances.

 Pharmaceutical preparation according to one of the preceding claims, characterised in that the propellant is TG 134a and/or TG 227.

20 4. Pharmaceutical preparation according claim 3, characterised in that the co-solvent comprises one or more pharmacologically tolerable alcohols, a pharmacologically tolerable ester, water or a mixture thereof.

 Pharmaceutcal preparation according to claim 3, characterised in that the co-solvent is ethanol.

- 6. Pharmaceutical preparation according to claim 3, 4 or 5 characterised in that the co-solvent is present in a concentration of up to 25% by weight, based on the quantity of liquefied propellant.
- 7. Pharmaceutical preparation according to claim 3, 4, 5 or 6, characterised in that the co-solvent is present

in a concentration of up to 10% by weight, based on the quantity of liquefied propellant.

- 8. Pharmaceutical preparation according to claim 1, 2,3,
 5. 4, 5, 6 and 7, characterised in that the composition is stabilised by a stabiliser.
- Pharmaceutical preparation according to claim 8,
 characterised in that the stabiliser contains one or more acids and/or salts.
- 10. Pharmaceutical preparation according to claim 8 or 9, characterised in that the stabiliser(s) contain(s)

 15 hydrochloric acid, sulphuric acid, nitric acid, phosphoric acid, ascorbic acid, citric acid, benzalkonium chloride and/or ethylene diamine tetraacetic and/or a salt thereof.
- 20 11. Pharmaceutical preparation according to claim 8 to 10, characterised in that the stabiliser is citric acid.
- pharmaceutical preparation according to one of the preceding claims 8 to 11, characterised in that the stabiliser is present in an amount of up to 100 ppm.
- 13. Pharmaceutical preparation according to one of the preceding claims 8 to 11, characterised in that the stabiliser is present in an amount of up to 40 ppm.
 - 14. Pharmaceutical preparation according to one of the preceding claims 1 to 13, characterised in that the preparation contains a surfactant or suspensionstabilising agent.

- 15. Pharmaceutical preparation according to claim 14, characterised in that the surfactant contains C_{5-20} fatty alcohols, C_{5-20} fatty acids, C_{5-20} fatty acid esters, lecithin, glycerides, propyleneglycol esters, polyoxyethanes, polysorbates, sorbitan esters and/or carbohydrates.
- 16. Pharmaceutical preparation according to claim 14, characterised in that the surfactant contains C_{5-20} fatty acids and/or the esters thereof.
 - 17. Pharmaceutical preparation according to claim 14, characterised in that the surfactant contains oleic acid and/or sorbitan mono-, di- or trioleate.

15

- 18. Pharmaceutical preparation according to claim 14, characterised in that the surfactant contains oleic acid.
- 20 19. Pharmaceutical composition according to claim 14 characterised in that the surfactant or suspension-stabilising agent comprises a toxicologically acceptable polymer and/or block-polymer.
- 25 20. Pharmaceutical preparation according to one of the preceding claims, characterised in that the active substance combination contains beclomethasone, budesonide, cromoglycinic acid, fenoterol, flunisolide, fluticasone, ipratropium, nedocromil, orciprenaline, oxitropium bromide, reproterol, salbutamol, salmeterol (albuterol), terbutalin, N-[[2,2-dimethyl-4-(2-oxo-2H-pyridin-1-yl)-6-trifluoromethyl-2H-1-benzopyran-3-yl]methyl]-N-hydroxy-acetamide, the esters, salts and/or solvates thereof.

21. Pharmaceutical preparation according to one of claims 1 to 20, characterised in that the active substance combination contains salbutamol sulphate (albuterol sulphate) and ipratropium bromide.

Intel anal Application No PCT/US 99/12785

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 A61K9/12 A61K A61K31/135 A61K31/46 According to International Patent Classification (IPC) or to both national classification and IPC Minimum documentation searched (classification system followed by classification symbols) IPC 6 A61K Documentation searched other than minimum documentation to the extent that such documents are included in the folids searched Electronic data base consulted during the international search (name of data base and, where practical, soarch terms used) C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. WO 98 01147 A (RHONE POULENC RORER LTD 1-7.14 ;BELL ALEXANDER (GB)) 15 January 1998 (1998-01-15) page 2, line 22-25 20,21 page 3, line 1-23 page 4, line 17 -page 5, line 8 examples 3.4 claims 1-4,6,12,20 1.3 X US 5 589 156 A (HENRY RICHARD A) 31 December 1996 (1996-12-31) abstract column 3, line 13-31 column 7, line 35-60 claims -/--Patent family members are listed in annex. Further documents are listed in the continuation of box C. Special categories of cited documents : T later document published after the international filing date or priority date and not in conflict with the application but "A" document defining the general state of the art which is not considered to be of particular relevance cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "E" earlier document but published on or after the International filmo date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another clietion or other special reason (as especified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such docu-"O" document referring to an oral disclosure, use, exhibition or ments, such combination being obvious to a person skilled in the art. document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the ectual completion of the international search Date of mailing of the international search report 28 September 1999 04/10/1999 Name and mating address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040. Tx. 31 651 epo nl. La Gaetana, R Fax: (+31-70) 340-3018

Inter unal Application No PCT/US 99/12785

		/05 99/12/65
	ation) OOCUMENTS CONSIDERED TO BE RELEVANT	D-I
Category °	Citation of document, with Indication, where appropriate, of the relevant passages	Relevant to daim No.
γ .	US 5 603 918 A (MCNAMARA DANIEL) 18 February 1997 (1997-02-18) column 2, line 31-59 examples claims	20,21
A	WO 94 13262 A (JAGER PAUL D ; KONTNY MARK J (US); NAGEL JURGEN H (DE)) 23 June 1994 (1994-06-23) page 4, line 12-21 page 5, line 22,23 page 6, line 1-7 page 8, line 3-6 page 10, line 6-11 page 10, line 17-30 page 16 -page 18; tables 2-4 claims 1-3,6,8-10,12,32	1,3-6, 8-18,20, 21
A	EP 0 499 344 A (RIKER LABORATORIES INC) 19 August 1992 (1992-08-19) page 2, line 39-50 page 4, line 36 -page 5, line 5 page 6, line 1-10 examples 4-6,24 claims 1,2,6,10,12	1,3-6, 14-21

information on patent family members

Inter anal Application No PCT/US 99/12785

	ent document		Publication		atent family	Publication
cited	in search report		date		nember(s)	date
MU	9801147	Α	15-01-1998	AU	3453897 A	02-02-1998
		••		ĔΡ	0914143 A	12-05-1999
			~			
US	5589156	Α	31-12-1996	US	5534242 A	09-07-1996
				US	5453445 A	26-09-1995
				US	5593661 A	14-01-1997
				AU	4711396 A	08-10-1996
				CA	2215680 A	26-09-1996
				WO EP	9629066 A 0814793 A	26-09-1996 07-01-1998
				JP	11502202 T	23-02-1999
				US	5858331 A	12-01-1999
				AŬ	4711496 A	16-10-1996
	•			CA	2215711 A	03-10-1996
				WO	9630008 A	03-10-1996
				EP	0814794 A	07-01-1998
				JP	11502526 T	02-03-1996
				US	5679325 A	21-10-1997
US	5603918	A	18-02-1997	NONE		
WO	9413262	A	23-06-1994	AT	177941 T	15-04-1999
	- 104646	••	VV 0447	ÄÜ	680227 B	24-07-1997
				AU	5740594 A	04-07-1994
				AU	6048694 A	04-07-1994
			BG	99760 A	29-02-1996	
				CZ	9501490 A	13-12-1995
				DE	69324161 D	29-04-1999
				EP	0673240 A	27-09-1995 01-06-1999
				ES FI	2129117 T 952842 A	09-06-1995
				GB	2288978 A,B	08-11-1995
				HU	72985 A	28-06-1996
				JP	8509459 T	08-10-1996
				LV	10911 A	20-12-1995
				LV	10911 B	20-04-1996
				NO	952269 A	08-06-1995
			•	NZ	259192 A	26-05-1997
				PL .	309333 A	02-10-1995
				SG SK	52459 A 76095 A	28-09-1998 08-01-1997
			MO 2 <i>Y</i>	76095 A 9413263 A	23-06-1994	
				US	5676930 A	14-10-1997
				CN	1095265 A	23-11-1994
				ZA	9309195 A	08-06-1995
	0400244		19-08-1992		6211EE D	19-11-1992
Ľ٢	0499344	A	12-00-1224	AU Au	631155 B 4595689 A	14-06-1990
			CA	2004598 A	06-06-1990	
			DE	68924540 D	16-11-1995	
			DE	68924540 T	15-05-1996	
		•	DK	595789 A	07-06-1990	
				EP	0372777 A	13-06-1990
				EP	0653204 A	17-05-1995
				ES	2045470 T	16-01-1994
				ES HK	2077971 T 7196 A	01-12-1995 26-01-1996
				P1 K	/ 190 A	てい_ハブニブシカの
				HK	80497 A	20-06-1997

...comation on patent family members

Interr nat Application No
PCT/US 99/12785

Patent document Publication cited in search report date		Publication date		atent family member(s)	Publication date	
EP 0499344	A		IL	92457 A	21-10-1994	
			JP	2200627 A	08-08-1990	
			JP	2786493 B	13-08-1998	
		•	MX	9203480 A.B	01-07-1992	
			US	5695743 A	09-12-1997	
			ŪŠ	5683677 A	04-11-1997	
			ÜS	5674473 A	07-10-1997	
			ÜS	5225183 A	06-07-1993	
			ÜS	5766573 A	16-06-1998	
			ÜS	5776434 A	07-07-1998	
			ÜS	5720940 A	24-02-1998	